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(54) Fibre containing composition

(57) Compositions of a water in water emulsion having a first phase of gelled particles with specific aspect ratio and width, are suitable for application in products. Said compositions impart a creamy impression to the final product.

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Description

Field of the invention

[0001] The invention relates to a process for producing a composition comprising two phases and to the product thus obtained, and to food products containing said composition.

Background to the invention

[0002] It is an object of the current invention to provide a product, especially a food product that shows a thick consistency even at relatively high shear rates.

High shear rates are encountered when applying a spread like composition to bread. Examples of such composition are fresh cheese and low fat spreads. [0003] The invention especially relates to food prod-

ucts showing a creamy mouthfeel.

Products giving a creamy mouthfeel are known in the 20 art. Example of such products are dairy creams, spreadable dairy products such as those disclosed in EP-A-841956, full fat fresh cheese type products, creme fraiche and like products.

[0004] These products impart a creamy mouthfeel 25 even at the high shear rates of chewing the product. In general such compositions are largely based on fat and/ or getatine or getatine replacers.

[0005] It is known to reduce the fat content of such products by including biopolymers as fat replacers. So However the resulting products usually do not impart creamy mouthled in the way the original products do. [0006] EPA-432835 discloses fluid compositions comprising attest one chemically setting gelling agent. The compositions can be obtained by shearing a liquid sonationing a chemically setting gelling agent, while gelation occurs. The compositions comprise gel particles having a mean diameter of preferably less than 100 micrometer. Such compositions altegedly possess storurable rhoological properties. The compositions obtained documprise a suspension of irregularly shaped gel particles that are effective at increasing the viscosity at low shear rates, but less so at this phear rates.

[0007] EPA-674,973 discloses the preparation of a composition ground at least one of the two phases is gelled. The processing of the two phases takes place in conventional A- and C units. These units impart to the composition turbulent flow conditions which lead to products with a 3d disporsed phase which has a variety of shapes which are mainly spherical with a broad average diameter distribution for the dispersed phase particles.

These compositions were found not to contribute to a creamy mouthfeel in the final products.

[0008] Furthermore Wolf et al disclose in Food hydrocolloids 14 (2000), 217-225 the effect of shear history on microstructure in biopolymer mixtures that form water-in-water emulsions upon de-mixing. It discloses the use of simple shear to manipulate phase morphology of water-in-water emulsions in order to produce regular elipsoidal or elongated structures. Cooling at the same time as applying shear is used to get the biopolymers and trap the structures in order to produce anisotropic get particles.

[0009] Furthermore Wolf et al disclose in Rheologica Acta 40 (2001), 238-247 the production of particles with 10 high aspect ratio. This document does not disclose nor suggest which compositions could be applied in food products. Furthermore this document does not disclose a continuous production process for such particles.

[0010] It is therefore an object of the current invention to provide a composition which when used in a product imparts a creamy impression upon use.

Summary of the invention

0 [0011] It has surprisingly been found that a composition comprising a first aqueous phase of gelled particles and a second aqueous phase, wherein the particles of the first phase are characterized by a specific aspect ratib in combination with a specific average width, imparts to products the desired creamy impression.

[0012] Therefore the invention relates to a composition comprising a first aqueous phase of gelled particles and a second aqueous phase wherein the particles of the first phase are characterized in that a minimum of 80 vol% of the particles have an aspect ratio of at least 2, and a maximum width of 100 µm and in that a minimum of 50 vol% of the particles have an aspect ratio of at least 5, and a maximum width of 40 µm and a minimum of 20 vol% of the particles have an aspect ratio of 5 at least 10, and a maximum width of 20 µm.

[0013] In a further aspect the invention relates to products comprising this composition and to a process for the preparation of these products.

40 Detailed description of the invention

[0014] The compositions according to the invention are capable of imparting creaminess to a product at surprisingly low concentrations. Furthermore the composition and products containing these are freeze thaw stable, and show de/re hydration stability. Advantageously the compositions according to the invention maintain a relatively high viscosity at this shear conditions.

The compositions known from prior art usually do not show this relatively high viscosity at high shear but show a greater degree of so called shear thinning behaviour under shear than the present composition. Without wishing to be bound by any theory it is believed that this behaviour occurs because the gelled particles adopt random orientations and configurations when shear is applied to the composition, that persist at high shear rates. This is not disclosed in any of the aforementioned publications, where the celled particles are aligned in

the flow direction and are different as a result.

[0015] In the description and claims where weight %

(wt%) is used this is weight% on total product weight unless otherwise indicated.

[0016] Aspect ratio is calculated by dividing the length of a fibre by the width.

The accuracy of the data relying on analysis of aspect ratio and width show a deviation of ± 10%. The determination of aspect ratio and width is defined in the examples. Both are well known parameters for characterisation of dispersed phase shape and size, see e.g. The Image Processing Handbook 2nd edition' by JC Russ, CRC Press London, ISBN 0-8493-2516-1, published in 1994.

[0017] The composition according to the invention is a dispersion comprising two aqueous phases. The first phase is a phase of gelled particles. Such particles are present dispersed in the second phase.

[0018] The gelled particles of the first phase are characterised by a specific aspect ratio in combination with a specific was peculiar aspect ratio of a specific with A minimum of 80 vol% of the particles have an aspect ratio of at least 2, and a maximum width of 100 µm and a minimum of 20 vol% of the particles have an aspect ratio of at least 5, and a maximum width of 40 µm and a minimum of 20 vol% of the particles have an aspect ratio of at least 10, and a maximum width of 20 µm. This distribution of particle aspect ratio and width can be considered a broad distribution. Without wishing to be bound by any theory, applicants believe that this broad distribution imparts to the compositions the de-sired behavior under shear and the limited shear thinning properties of the claimed material.

[0019] Preferably at least 5 vol % of the particles are characterised by an aspect ratio of at least 50, and a maximum width of 20 µm. More preferably the particles are characterised by a minimum width of 0.5 µm, more preferred 1 µm.

[0020] The particles of the first phase are gelled particles. Gelled particles are defined as particles that may have shapes ranging from spherical to ellipsoidal to highly elongated in one direction, that comprise a gel forming compound solution in water which compound is in the gelled state.

[0021] The gelled particles preferably comprise a biopolymer or a combination of biopolymers. Gelling bipolymers are generally known and include for example gellan, k-carrageenan, sodium alginate, gelatine, agar, agarose, maltodextrin, and heat set proteins.

[0022] Gelation can be obtained in any suitable way.
The gelation freatment is preferably selected from the 50 group comprising a temperature treatment, chemical gelation or crystallisation. The gelation method that is selected depends on the ingredient composition of the dispersed phase and of the continuous phase and of the continuous phase.

[0023] Gelation by temperature treatment is selected 55 if a gelling agent is used, whose setting is dependent on temperature. Examples of such gelling agents Include gelatine, which sets at a temperature of below about 40

*C and agar which sets at a temperature of below about 45 *C and carrageman or gellan whose gelstion temperatures are dependent on sait type and concentration (reference is made to Handbook of hydrocolloids, edited by G.O. Phillips and P.A. Williams, published by GRO. Press). Also proteins that gelform a network by heat treatment are suitable for preparation of the two phase composition.

[0024] It will be appreciated that the exact gelling temoperature for the biopolymer used is determined, among others, by quality, purity, concentration, solvent properties such as added salt or sugar, and pH. [0025] According to another embodiment, a chemical-

by setting gelling agent is used. By a chemically setting 5 gelling agent is meant a component which, after being dispersed in another phase such as a liquid, will set to a gel when allowed to chemically interact with a supplementary active component, whose active component is usually a cation, or which sets due to the occurrence of a chemical reaction such as oxidation. A gelling agent setting upon a pH change is also encompassed in the term chemically setting gelling agent. Examples of such pH dependent gelling agents are proteins which will generally set or precipitate at a pH below the iso-electric 5 point.

[0026] In such cases where a chemically setting gelling agent is applied, chemical gelation is preferably applied.

Chemical gelation can be obtained by combining the gelling agent with a salt with an effective callon to form a salt of the gelling agent and the cation. The combination of the gelling agent with the cation may be effected by the addition of the cation as such or alternatively by converting a procursor compound, present in the phase comprising the gelling agent, or the other phase, into the free, effective, cation. The cation is preferably selected from Ca²⁺ and K⁺, Na⁺ and mixtures thereof, the most preferred total is Ca²⁺.

[0027] In an even more preferred embodiment, the gelling agent is selected from the group comprising x-carrageenan, pectin, iota-carrageenan, hrorelleran, carboxymethyl cellulose, gellan, gelatine, alginate, agar, guar or a combination thereof; most preferably gellan or x-carrageenan.

45 [0028] In a preferred embodiment gelation by temporature treatment is used. The most preferrable temperature treatment is used. The most preferrable temperature treatment is cooling, hence, a biopolymer gelling upon cooling is most preferably used in the first phase. [0029] The amount of first phase compared to second op phase among others depends on the application that is envisaged. In general it is preferred that the phase volume of the first, dispersed, phase in the second, continuous, phase is from 10 to 40 vol%.

[0030] As described above, it is required that at least the first phase is gelling.

The second phase is preferably composed such that the interfacial tension between the two phases in their liquid state and on onset of gelation of the first phase is suffi-

ciently low to prevent the particles of the first phase from hashe relaxation during or after gelation. By shape relaxation we mean significant if not total loss of anisotropic particles hashed. The properties of the properties

[0031] According to a preferred embodiment, the first phase comprises a gelling biopolymer, and the second phase comprises a non-celling biopolymer.

[0032] According to another embodiment, both the first and the second phase comprise a gelling biopolymer.

[0033] In a preferred embodiment the biopolymer combination for the first and second phase is selected from the following combinations

- a) gelatine first phase quar second phase
- b) gellan first phase κ-carrageenan second phase c) gellan first phase - sodium alginate second phase d) κ-carrageenan first phase - carboxymethylcellulose second phase.

[0034] Furthermore the compositions can be made heat stable by taking some further straightforward measurements such as increasing the ion content, adding gef stabilisers or using heat set proteins.

[0035] The composition can suitably be applied in any type of water continuous product. Examples of products are food products, preferably dairy type spreads, dressings, sauces, or frozen desserts.

[0036] The amount of composition that can be applied 40 varies depending on the specific purpose and final product.

In a preferred embodiment, the amount of composition according to the invention contained in a final product is at most 30 w%. Higher amounts could lead to products which are too thick for application. Preferred products comprise from 3 to 20 wt% of the composition according to the invention.

[0037] The composition is especially suitable for Inclusion in water continuous food products such as dairy
type spreads. In a preferred embodiment the invention
relates to a spreadable dairy type product comporting
from 5 to 35 wf% fat, 0.2 to 10 wf% protein, and from 3
to 20 wf% of a composition comprising a first aqueous
phase of gelled particles and a second aqueous phase,
wherein the particles of the first phase are in the form of
elongated fibers.

[0038] More preferably the invention relates to a

spreadable dairy type product comprising from 5 to 35 wt% fat, 0.2 to 10 wt% protein, and from 3 to 20 wt% of a composition comprising a first aqueous phase of gelled particles and a second aqueous phase, wherein

- 5 the particles of the first phase are characterised in that a minimum of 80 v0% of the particles have an aspect ratio of at least 2, and a maximum width of 100 µm and in that a minimum of 50 v0% of the particles have an aspect ratio of at least 5, and a maximum width of 40 µm and a minimum of 20 v0% of the particles have despect ratio of at least 10, and a maximum width of 20 v0% of the particles have for the spect ratio of at least 10, and a maximum width of 20 v0% of the particles have for the spect ratio of at least 10, and a maximum width of 20 v0% of the particles have for the spect ratio of at least 10, and a maximum width of 20 v0% of the particles have for the spect ratio of at least 10, and a maximum width of 20 v0% of the particles have for the spect ratio of the spect of the spect ratio of the spect of the spect ratio of t
- [0039] Preferably the composition included in such dairy type spread comprises a first phase comprising 5 gelled x-carrageenan and a second phase comprising carbox methylicallulose.

[0040] Optionally such products further include a salt, other biopolymers, preservation agents, flavourings or other food grade ingredients.

[0041] Typical compositions suitable for a dairy type spread are e.g. disclosed in WO-A-97/04860 and WO-A-97/08956. These products are generally acidified creams whereby the cream may be based on dairy fat, vegetable fat

²⁵ [0042] The composition according to the invention may be prepared by any suitable process. However the process below is highly preferred as it enables preparation of the composition in a simple, straight forward and economical efficient process. For example the process introduced in Rheologica Acta 40 (2001), 238-247 discloses a process which is unpreferred for preparation of said composition as this process is unsuitable for appli-

[0043] Therefore in a further embodiment, the invention relates to a process for the preparation of a composition comprising a first aqueous phase of gelled particles and a second aqueous phase, said process comprising the steps of

cation on large, industrial scale.

 a) mixing two aqueous phases each comprising a polymer, preferably a biopolymer, wherein at least one of the polymers is a gelling biopolymer,

 b) treating the mixture according to (a) such that one of the phases is present in the form of droplets in the second phase,

c) subjecting the mixture to shear flow,

 d) subjecting the mixture to a gelation treatment during or after step (c);

wherein step (a) and (b) are carried out in a pre-mix tank and step (c/d) are carried out in a cylindrical pipe, or an array of cylindrical pipes.

[0044] In the context of the invention shear flow is defined as planar flow as shown in figure 1.

5 Turbulent flow is unsuitable for the process according to the invention. This type of flow is for example encountered in a scraped surface heat exchanger (A-unit), a pin stirrer (C-unit), a homogeniser and rotor-stator based systems which are well known equipment for preparation of emulsions. Documents wherein these types of apparatus are applied are for example US-A-5,659,000, EP-A-432,835, US-A-6,165.534 and WO-A-99/51716. The turbulent flow in these apparatus' is ar-5

bitrary and undefined.

10045] It is preferred that the gelled aqueous phase particles are in the form of elongated fibres, preferably characterised in that a minimum of 80 vol% of the particles have an aspect ratio of at least 2, and a maximum width of 100 µm and in that a minimum of 50 vol% of the particles have an aspect ratio of at least 5, and a maximum width of 40 µm and a minimum of 20 vol% of the particles have an aspect ratio of at least 10, and a maximum width of 40 µm and a minimum of 20 vol% of the particles have an aspect ratio of at least 10, and a maximum width of 20 µm.

[0046] In a preferred embodiment, step (a) is simplified by mixing the two biopolymers as dry powders, and adding them in one step to a common aqueous selvent. Depending on the biopolymer combinations used, the mixture of dry powder additionally comprises salt such 20 as sodium chloride or potassium chloride in a preferred concentration. Sugar can also be added to improve solvent conditions.

[0047] It will be appreciated that the characteristics of the cylindrical pipe or array of pipes is preferably such that they are sufficiently long and sufficiently small in diameter so that gelation of the first phase occurs within the pipe.

[0048] In a preferred embodiment, the flow rate in a single cylindrical pipe is from 0.1 to 25 mLmirri - and the 30 wall shear stress is from 15 to 800 Pa. For the definition of wall shear stress reference is made to textbooks on rheology such as for example: Rheology: Principles, Measurements, and Applications, C. W. Macosko, VCH Publishers, Inc., 1994.

[0049] The gelation can be obtained in any of the ways described above.

[0050] The final product comprising the composition according to the invention can be prepared in any suitable way. In the following, the composition according to 40 the invention is denoted as fibre composite or fibre composite material.

[0051] According to one embodiment a base product is prepared from all ingredients except for the fibre composite material. Ingredients are for example proteins, 45 polysaccharides, fat or oil, salt, sugar, flavours, acids, or other ingredients normally found in such products, and water. Processing can include mixing, homogenisation, emulsification, whipping, heating, and cooling. The fibre composite material is preferably then added to the 50 base product while stirring, sometimes at elevated temperature. Stirring preferably continues till the product appears to be a homogenous mixture of all ingredients, in particular the fibre composite. In some cases the product is then filled directly into product containers, and 55 stored under appropriate conditions. This applies to the preparation of dressings and sauces. For the preparation of dairy type spreads an acidification step preferably

follows prior to filling. When a frozen desert product, e. g. ice cream, is prepared, after incorporation of the fibre composite material, the product is whipped, and frozen according to a conventional house hold or factory ice cream process. The final product is stored in a freezer at preferably ~18°C.

[0052] According to an alternative embodiment the composition according to the invention is submittled to a drying step and in that form added to the final product of at any stage of the process. Drying techniques such as freeze drying, or vacuum drying were found to be suitable. Alternatively, the composition according to the invention can be forzed for storage, and added in its frosers and the process of the product ensuring that the composition have scompletely to impart the desired croperties on the

[0053] According to another alternative embodiment, the composition according to the invention is prepared in situ during the preparation of a final product in which the composition is included.

[0054] In a further aspect the invention relates to an apparatus suitable for carrying out the process described above, comprising a pre-mix tank and a cylindrical pipe comprising means for obtaining gelation.

[0055] The invention is now illustrated by the following non-limiting examples.

Examples

product.

General

[0056] Oil droplet size is determined by use of well known static laser diffraction.
[0057] The procedure of particle shape analysis is described in the following. The fibre composite material

was diluted with an aqueous solvent of the same ionic strength as present in the composition. This means the fibre composite was diluted with its own solvent. For dilution a certain amount of the fibre composite was added to a certain amount of the solvent while stirring with a paddle stirrer, or shaking in a flask rotator to prepare a diluted sample for particle shape analysis that contained roughly 1 wt% fibres. In case of a gelled second phase, the type of salt to get the desired ionic strength was carefully chosen in order to melt the gel of the second phase on dilution ensuring the particles of the first phase remained gelled. As a next step, images of the sample were taken using conventional light microscopes under suitable lighting conditions preferably phase contrast. and magnification as low as possible preferably using a 10X phase contrast lens and setting any further means to influence magnification at the lowest setting such as 0.8 on a magnification wheel when fitted to the microscope used. The images taken here pictured a real length of 800 um along their width. The width of a fibre was measured at least once, most preferably several times along the fibre length at equally spaced locations

which were roughly a tenth of the image width apart.

hence since the width of the images was 800 um, the fibre width was measured every ~ 100 µm along the fibre length. Typically, images showed a number of separate fibres, very small apparently spherical shaped particles, and larger entities that appeared to be associated 5 fibres. Enough images were statistically taken such that 400 fibres were pictured as individual entities meaning they don't overlap with any other fibres. A particle was identified as a fibre when it showed a clear extension in one direction. If a fibre was not pictured with its whole length, the pictured length of this fibre was measured and taken for aspect ratio calculation. The aspect ratio was calculated by dividing the length of a fibre by the width. The width of a fibre was measured at least once. most preferably several times along the fibre length at 15 equally spaced locations which were roughly a tenth of the Image width apart, e.g., a magnification of 1:40 corresponds to an image width of 800 µm, hence, the fibre width was measured every ~ 100 µm along the fibre length. Finally, the average of the measured values was 20 taken as the width of the fibre.

Example 1

[0058] A fibre composite was prepared from a mixture $\,\,\,$ 25 of $\kappa\text{-}carrageenan$ and CMC with the following material characteristics.

k-carrageenan:

[0059]

- A speciality κ-carrageenan (Genuvisco X0909, ex Hercules Limited (UK)) was used, with low residual cations, in particular low Potassium.
- The typical composition on powder is:

92.9 wt% κ-carrageenan 5.12 wt% Sodium ions 0.18 wt% Potassium ions 0.01 wt% Calcium ions

- 0.01 % w/w Magnesium ions
- The moisture content of the powder is 10 wt%.
 The zero shear viscosity of a 1% w/w (on powder) 45 aqueous solution at 20°C is 0.08 Pa.s ± 10%.

carboxymethylcellulose (CMC):

100601

- A commercially available CMC (Blanose 7MF, ex CPKelco UK Limited) was used.
- The ion content of the batch used is:

7.6 wt% Sodium ions 7 10⁻³ wt% Potassium ions 11.8 10⁻³ wt% Calcium ions 22.6 10⁻³ wt% Choride ions 5 10⁻³ wt% Sulfate ions

- The moisture content of the powder is 5 wt%.
- The zero shear viscosity of a 1 wt% (on powder) aqueous solution at 20°C is 0.06 Pa.s ± 10%.

[0061] 60 g of said k-carrageenan, 150 g of said CMC. and 15 g Potassium Chloride were mixed dry, and then added to 2775 g de-ionised water while stirring vigorously with a paddle stirrer. The mixture was then heated up to 95°C under continued stirring, and kept under these conditions till a liquid mixture of all ingredients had formed. The mixture was transferred into a jacketed, and agitated pre-mix tank which was pre-heated to 95°C. The vessel could take a maximum of 5 l. In the pre-mix tank, the mixture was continued to be stirred. and after 15 min at 95°C the temperature was lowered to 80°C. Once this temperature was reached, the vessel was set under a hydrostatic pressure of 2.5 bar, and the outlet valve at the bottom of the pre-mix tank was opened to allow for the mixture to flow through a pipe with circular cross section which was jacketed for most of its length. The length and inner diameter of the pipe was 1.2 m and 8 mm respectively. The middle 0.5 m of the pipe were cooled by a water tacket, the temperature of the water incoming at the further end of the jacket was 10°C. The water exited at the end of the jacket which was closer to the pre-mix tank. The product was collect-30 ed under sterile conditions, and either directly used for preparation of a food product, dried, or stored in a freezer at -18°C. The shape characteristics of the product were not altered by the drying process, or the freezing process, and subsequent re-hydration or thawing respectively.

The hydrostatic pressure of 2.5 bar applied to the premix tank corresponds to a wall shear stress of 417 Pa. This fibre composite material comprises a non-gelled second aqueous phase dominated by CMC. The partidel phase comprises gelled k-carrageenan fibres. The volume in the mixture occupied by the first, gelled, phase is about 20%.

Example 2

[0062] Preparation of a fibre composite from gellan and κ-carrageenan.

gellan:

[0063]

50

- A commercially available gellan gum (Kelco gel F, ex CPKelco UK Limited) was used.
- the ion content is:

3.8 wt% Potassium lons 0.6 wt% Sodium ions

0.3 10⁻³ wt% Calcium ions

- The moisture content of the powder is 10 wt%.
- The zero shear viscosity of a 1 wt% (on powder) aqueous solution at 60°C (the solution forms a gel at 20°C) is 7.7 10-3 Pa.s ± 10%.

k-carrageenan:

[0064] The same x-carrageenan as specified in ex- 10 ample 1 was used.

[0065] The preparation of the fibre composite material from the two polysaccharides gellan and κ-carrageenan corresponds in its main steps to the previously described example 1. Differences were only in the quantities, the choice of temperatures, and pressure which are quantified below.

Quantities used were 60 g gellan, 60 g k-carrageenan and 2880 a de-ionised water. The temperature of the water feeding the jacket of the circular pipe was 5°C. 20 The hydrostatic pressure applied to the pre-mix vessel and the wall shear stress were 1.1 bar and 183 Pa re-

This fibre composite material comprises a gelled second aqueous phase which is dominated by x-carrageenan. 25 The first gelled aqueous phase, the fibres, consist of 4.4 wt% gelled gellan. 32 %v/v of the fibre composite are occupied by the gelled gellan particle phase.

Example 3

Preparation of a dairy spread comprising gelled kcarrageenan fibres.

[0066] 1000 g dairy spread comprising 250 g fibre 35 composite material prepared from the polysaccharides κ-carrageenan and CMC as described in example 1, 3 g guar, 5 g salt (potassium chloride), 1 g potassium sorbate, 70 g buttermilk powder, 17.5 g whey protein, 220 g fat blend (1:1 fractionated coconut oil:fractionated palm oil), and 433.5 g water was prepared. The fibre composite material was previously frozen, and thawed prior to use. From the dry ingredients and the water a homogeneous mixture was prepared, which was then heated to 85°C at which temperature the melted fat was added 45 under stirring to from a coarse oil-in-water emulsion, In a next step, for decreasing the average size of the oil droplets to ca. 1 um, a conventional emulsification step using a high pressure homogeniser followed. The resulting product base was cooled to 40°C under continued 50 stirring with a paddle stirrer, and the fibre composite material was added in portions of roughly 10 g. After a mixing time of 15 minutes the product was acidified to a pH of 4.7 with 80% lactic acid. This concluded the production of the dairy spread with x-carrageenan fibres, and 55 the product underwent a sterile filling process into sample containers, and was stored at 6°C.

100671 The resulting composition was found to be

creamy upon consumption.

Additionally, the resulting composition was described as having a long texture upon spooning compared to commercial dairy spread products Brunchtm and Creme Boniourtm.

Example 4

Preparation of a fresh cheese comprising gelled gellan fibros

[0068] 1000 g fresh cheese comprising 156.2 g fibre composite material prepared from the polysaccharides gellan and k-carrageenan as described in example 2.3 g guar, 3 g salt (sodium chloride), 1 g potassium sorbate, 70 g buttermilk powder, 17.5 g whey protein, 220 g fat blend (1:1 coconut oil:palm oil), and 529,25 g water was prepared. The preparation of the fresh cheese followed the same route as the preparation of the dairy spread described in example 3. The only difference was a temperature of 60°C (instead of 40°C) at which the fibre composite was mixed into the product base.

[0069] The resulting composition was found to be creamy upon consumption. Additionally, the resulting composition was found to be

smooth, and mouth coating.

Example 5

Preparation of an ice cream comprising gelled kcarrageenan fibres

[0070] An ice cream comprising 25 wt% fibre composite material prepared from the polysaccharides κ-carrageenan and CMC as described in example 1, 8 wt% butterfat, 10 % w/w skim milk powder, 0.3 wt% emulsifier (monoglycerolpalmitate), 13 wt% sucrose, 4 wt% glucose syrup (malto dextran with a DE of 40), 0.012 wt% vanillin, 0.016 wt% carrageenan L100 (which is a blend of 30wt% sucrose, 34.3 wt% k-carrageenan, 4.2 wt% tcarrageenan, 31.5 wt% λ-carrageenan), 0.144 wt% iocust bean gum, and 39.528 wt% water was prepared. The dry ingredients were mixed together, and then added to the water which was heated up to 40°C. A homogeneous mixture was prepared by stirring the mixture, then the butterfat was added, and the mixture heated to 82°C under continued stirring. After a further 5 minutes of stirring at 82°C the mixture was homogenised using a standard process in ice cream manufacturing process (high pressure homogeniser, fitted with two tapered valves, 140 bar or 2000 psi homogenisation pressure, pre-warmed with hot water). The base product was then cooled to 40°C where the fibre composite was added under stirring with a paddle stirrer till a homogeneous mixture had formed. For subsequent aeration and freezing, a bench top whisker (hobart ex Hobart UK) and a house hold ice cream maker (model gelato chef 2000 ex magimix UK LTD) respectively were used. The residence time of the product in the whisker was 15 minutes. Overruns of 30% were achieved. In the house hold ico cream maker the product was cooled down to -2°C, the product was then placed for 2 hours in a blast freezer at -37.5°C. For storage the product was kept in a house hold freezer at -13°C.

[0071] The product was tasted after it reached a homogeneous temperature of -18°C. It was described as creamy, and rich in taste.

Claims

- 1. Composition comprising a first aqueous phase of gelled particles and a second aqueous phase, 15 wherein the particles of the first phase are characterised in that a minimum of 80 vol% of the particles have an aspect ratio of at least 2, and a maximum width of 100 µm and in that a minimum of 50 vol% of the particles have an aspect ratio of at least 2, and a maximum width of 40 µm and a minimum of 20 vol% of the particles have an aspect ratio of at least 10, and a maximum width of 20 µm.
- Composition according to claim 1 wherein at least 5 vol % of the particles are characterised by an aspect ratio of at least 50, and a maximum width of 20 μm.
- Composition according to claim 1 wherein the particles are characterized by a minimum width of 0.5 μm.
- Composition according to claim 1, wherein the dispersed phase comprises a gelling biopolymer.
- Composition according to claim 1 wherein the phase volume of the first, dispersed, phase in the second, continuous, phase is from 10 to 40%.
- Composition according to any of claims 1-5 wherein
 the biopolymer is selected from the group comprising x-carrageenan, pectin, iota-carrageenan, furcelleran, carboxymethyl cellulose, agar, gellan, gelatine, alginate, guar or a combination thereof.
- Product comprising the composition according to any of claims 1-6 in an amount of at most 30 wt%.
- Product according to claim 7 comprising from 3 to 50 20 wt% of the composition according to claim 1.
- Spreadable dairy type product comprising from 5 to 35 wt% fat, 0.2 to 10 wt% protein, and from 3 to 20 wt% of a composition comprising a first aqueous phase of gelled particles and a second aqueous phase, wherein the particles of the first phase are in the form of elongated fibres.

- Spreadable dairy type product comprising from 5 to 35 wt% fat, 0.2 to 10 wt% protein, and from 3 to 20 wt% of a composition according to claim 1.
- i 11. Spreadable dairy type product according to claim 9 or claim 10, wherein the composition comprises a first phase comprising gelled x-carrageenan and a second phase comprising carboxymethy(cellulose.
- 10 12. Process for the preparation of a composition comprising a first aqueous phase of gelled particles and a second aqueous phase, said process comprising the steps of
 - a) mixing two aqueous phases each comprising a polymer, preferably a biopolymer, wherein at least one of the polymers is a gelling polymer, b) treating the mixture according to (a) such that one of the phases is present in the form of droplets in the second phase,
 - c) subjecting the mixture to shear flow,
 d) subjecting the mixture to a gelation treatment during or after step (c)
 - and wherein step (a) and (b) are carried out in a premix tank and step (c/d) are carried out in a cylindrical pipe, or an array of parallel cylindrical pipes.
 - 13. Process according to claim 12 wherein the flow rate in a single cylindrical pipe is from 0.1 to 25 ml.min⁻¹ and the wall shear stress is from 15 to 800 Pa.
 - Apparatus suitable for carrying out the process according to claim 12 comprising a premix tank and a cylindrical pipe comprising means for obtaining gelation.

Fig.1.

Type of flow

Simple shear



Flow pattern

Drop shape

B

 $\mathrm{d}u_Z/\mathrm{d}y$

Velocity gradient 6=

Rotation rate

G/2



PARTIAL EUROPEAN SEARCH REPORT

Application Number

which under Rule 45 of the European Patent ConventionEP 02 07 7857 shall be considered, for the purposes of subsequent proceedings, as the European search report

		ERED TO BE RELEVANT			
Category	Citation of document with of relevant pas	indication, where appropriate, sages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.CI.7)	
X	(NL)) 22 December : * page 2, line 48 -	[LEVER PLC ;UNILEVER NV 1993 (1993-12-22) - page 3, line 19 * page 5, line 11; claims	1-14	A23L1/05 A23D7/00	
x	ET AL) 24 October 2 * column 1, line 59	9 - column 2, line 44 * 0 - column 4, line 6;	1-14		
x	US 5 952 007 A (VIS 14 September 1999 (* column 2, line 39 claims 1,2,5-7 *		1-14		
		-/			
				TECHNICAL FIELDS SEARCHED (Int.CL7)	
				A23L A230	
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Claims no	t searched :				
Reason fo	or the limitation of the search;				
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	Place of search	Order of complation of the search		Examiner	
	MUNICH	9 September 2002	Ver	nier, F	
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INCOMPLETE SEARCH SHEET C

Application Number EP 02 07 7857

Claim(s) searched incompletely:

Reason for the limitation of the search:

Present claims 1-14 relate to a composition/dairy type product/method/apparatus defined by reference to the following parameters:

- P1: maximum width
- P2: aspect ratio
- P3: vol%

The use of these parameters in the present context is considered to lead to a lack of clarity within the meaning of Article 84 FPC. It is impossible to compare the parameters the applicant has chosen to employ with what is set out in the prior art. The lack of clarity is such as to render a meaningful complete search impossible. Consequently, the search has been restricted to:

- a, composition of a gelled biopolymer in an aqueous solution containing another biopolymer
- b. spreadable dairy type product containing said composition
- c. process for obtaining said composition



PARTIAL EUROPEAN SEARCH REPORT

Application Number

EP 02 07 7857

	DOCUMENTS CONSIDERED TO BE RELEVANT	GLASSIFICATION OF THE APPLICATION (Int.CI.7)	
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	
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EP 02 07 7857

This annex lists the patent family members relating to the patent documents clied in the above-mentioned European search report. The members are as contained in the European Patent Office EDP file on The European Patent Office is in oway liable for these particulars which are merely given for the purpose of information.

09-09-2002 Patent document cited in search report Publication Patent family Publication date EP 0574973 22-12-1993 EΡ 0574973 A1 22-12-1993 AT 142428 T 15-09-1996 DE 69304615 D1 17-10-1996 69304615 T2 DE 20-02-1997 ΩK 574973 T3 24-02-1997 US 5464645 A 07-11-1995 HS 6136363 24-10-2000 ΑU 8804398 A Α 08-02-1999 9810684 A RR 22-08-2000 WO 9902047 A1 21-01-1999 EP 0994658 A1 26-04-2000 US 5952007 Α 14-09-1999 AT 180388 T 15-06-1999 684738 B2 AU 08-01-1998 5812194 A 19-07-1994 AU CA 2152386 A1 07-07-1994 CN 1090474 A 10~08-1994 CZ 9501638 A3 17-01-1996 DF 69325112 D1 01-07-1999 DE 69325112 T2 23-09-1999 DK 790780 T3 15-11-1999 NO 9414334 A1 07-07-1994 EP 0790780 A1 27-08-1997 2131671 T3 72131 A2 ES 01-08-1999 HÜ 28-03-1996 29-07-1994 MX 9400134 A1 PL 309632 Al 30-10-1995 SK 82995 A3 08-11-1995 9309417 A 15-06-1995

DPM.

For more details about this annex : see Official Journal of the European Patent Office, No. 12/82